

# OREL VINE SCHOOL

## Course Outline

### Cambridge IGCSE (Year 10)

#### CHEMISTRY

<i>Term I</i>		<b>Term II</b>		<b>Term III</b>	
<i>Topic/Subtopic</i>	<b>Contents</b>	<i>Topic/Subtopic</i>	<b>Contents/Objectives</b>	<i>Topic/Subtopic</i>	<b>Contents/Objectives</b>
<b><i>States of matter (4 hours)</i></b>	<ul style="list-style-type: none"><li>• Solids, liquids and gases</li><li>• Diffusion</li></ul>	<b><i>Chemical energetics (3 hours)</i></b>	<ul style="list-style-type: none"><li>• Exothermic and endothermic reactions</li></ul>	<b><i>Metals (4 hours)</i></b>	<ul style="list-style-type: none"><li>• Properties of metals</li><li>• Uses of metals</li><li>• Alloys and their properties</li><li>• Reactivity series</li><li>• Corrosion of metals</li><li>• Extraction of metals</li></ul>
<b><i>Atoms, elements and compounds (6 hours)</i></b>	<ul style="list-style-type: none"><li>• Elements, compounds and mixtures</li><li>• Atomic structure and the Periodic Table</li><li>• Isotopes</li><li>• Ions and ionic bonds</li><li>• Simple molecules and covalent bonds</li><li>• Giant covalent structures</li><li>• Metallic bonding</li></ul>	<b><i>Chemical reactions (5 hours)</i></b>	<ul style="list-style-type: none"><li>• Physical and chemical changes</li><li>• Rate of reaction</li><li>• Reversible reactions and equilibrium</li><li>• Redox</li></ul>	<b><i>Chemistry of the environment (4 hours)</i></b>	<ul style="list-style-type: none"><li>• Water</li><li>• Fertilisers</li><li>• Air quality and climate</li></ul>

<b><i>Stoichiometry</i></b> <b><i>(5 hours)</i></b>	<ul style="list-style-type: none"> <li>• Formulae</li> <li>• Relative masses of atoms and molecules</li> <li>• The mole and the Avogadro constant</li> </ul>	<b><i>Acids, bases and salts</i></b> <b><i>(4 hours)</i></b>	<ul style="list-style-type: none"> <li>• The characteristic properties of acids and bases</li> <li>• Oxides</li> <li>• Preparation of salts</li> </ul>	<b><i>Organic chemistry</i></b> <b><i>(8 hours)</i></b>	<ul style="list-style-type: none"> <li>• Formulae, functional groups and terminology</li> <li>• Naming organic compounds</li> <li>• Fuels</li> <li>• Alkanes</li> <li>• Alkenes</li> <li>• Alcohols</li> <li>• Carboxylic acids</li> <li>• Polymers</li> </ul>
<b><i>Electrochemistry</i></b> <b><i>(4 hours)</i></b>	<ul style="list-style-type: none"> <li>• Electrolysis</li> <li>• Hydrogen–oxygen fuel cells</li> </ul>	<b><i>The Periodic Table</i></b> <b><i>(4 hours)</i></b>	<ul style="list-style-type: none"> <li>• Arrangement of elements</li> <li>• Group I properties</li> <li>• Group VII properties</li> <li>• Transition elements</li> <li>• Noble gases</li> </ul>	<b><i>Experimental techniques and chemical analysis</i></b> <b><i>(4 hours)</i></b>	<ul style="list-style-type: none"> <li>• Experimental design</li> <li>• Acid–base titrations</li> <li>• Chromatography</li> <li>• Separation and purification</li> <li>• Identification of ions and gases</li> </ul>

## Resources

- Cambridge International IGCSE Biology Course book second edition
- Cambridge International IGCSE Chemistry course book
- Cambridge IGCSE Biology past papers
- Cambridge IGCSE Chemistry past papers
- Ezy Education e-learning platform

**Methodology**

- Discovery method
- Experimentation
- Field study
- Integration of Information technology (IT)
- Research work
- Use of primary and secondary data from various sources

**Mode of assessment**

- Daily exercises from the coursebook
- End of month tests
- End of unit tests set from Cambridge IGCSE Chemistry past papers
- End of unit tests set from Cambridge IGCSE Biology past papers
- Mid - term & end of term assessments
- Quizzes and assignments on the LMS
- Weekly homework assignments
- Ezy education e - learning platform.